

Silica Supported Copper Nano Particles Design from Metal Waste Apply Formation of Various Condensation Reaction

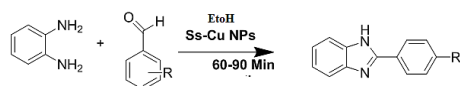
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Abstract. In recent years organic chemist focused on green chemistry for organic synthesis transformations because the silica supported catalyst is more stable, efficient and easy to recover and reusable after reaction. The silica supported copper catalyst is design from E-Waste, It used for formation of benzimidazoles from diamines and aldehydes.it is very deep work done by various catalyst, we noted in our research in unique method develop for synthesis of benzimidazoles made by condensation diamine and aldehydes via silica supported copper nano particles under EtOH as reaction media . The reactions done at 60° C temperature for 80 minute by 1:1.2 ratio mixture of diamine and aldehydes in the presence of nano catalyst to give the desired products with high purity..



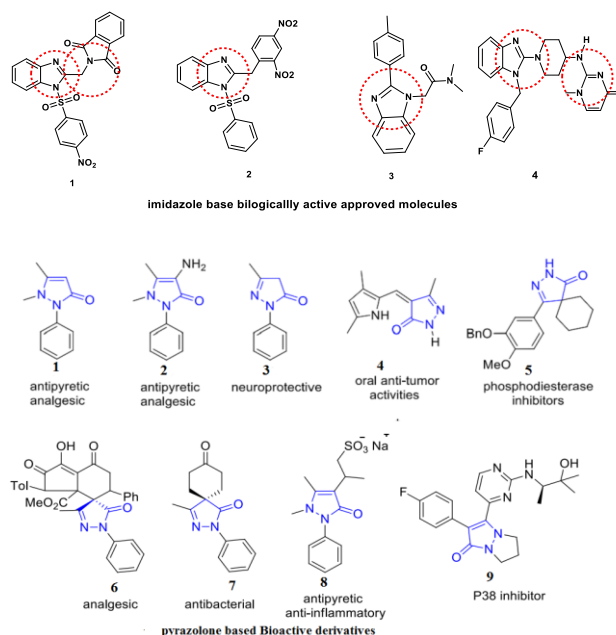
Index Terms: Scheme: 01

; Scheme 1. Synthesis of benzimidazole derivatives;

Waste management, green catalyst, high reusability, Biological evaluation

I. INTRODUCTION

Currently metal waste, plastic waste and e-waste is highest global environment issue ,House hold electronic gadgets PCB is made by copper wire, the gadgets are out of service its produced electronic waste, our team is collected this type house hold e-waste and design it's chemical process for formation of nano particles, it nano particles apply as a catalyst for formation of various know biologically active heterocyclic derivatives like benzimidazole, s via solvent free or aqueous media condition . Checked efficiency and reusability of catalyst. the all derivatives are based on the heterocyclic chemistry and highly applicable as lead molecules of biological activities of benzimidazole like anti HBV (1) [5], antitubercular (2) [6], non-seadating antihistamine(3) [7], (GABAA agonists [8], anti-viral (3) [9], antiinflammatory and analgesic [10], The drug discovery application of pyrazolone includes antipyretic activity (1,2,3,4) and analgesic activity (1,2,5,7,8,9),12,13,14,15 antibacterial ,16 anti-inflammatory activity antitumor activity (7,8,9),17,18 aneuroprotective and cardiovascular agent.



II. MATERIALS AND METHODS

A mixture of o-phenylenediamine (5 mmol) and sub.aldehyde (5.5 mmol) was stirred for 15 min than add novel designed Silica supported copper nano particles as catalyst at 28°C and use 5 ml Ethanol as a solvent at 60°C refluxed 80 to 90 minute and check TLC in Hexane: Ethyl acetate: Acetic acid(70:20:10) and reaction if complies, clear reaction mass filter by filter

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paper and remove catalyst than solvent remove from product and repurify in Ethanol and check its Melting point and TLC, sturture determination by IR by FTIR, ¹HNMR, ¹³CNMR and mass spectroscopy.

III. RESULTS AND DISCUSSION

We initiated our observation by using diamine (1.0 equiv.) and aromatic aldehyde (1.02 equiv.) in presence of silica supported copper nano particles (1-3%) in ethanol solvent at 55-65°C under reflux condition. the reaction goes out forward direction 80-85 % of

conversion and 10% KSM and 5% some unknown impurities , were isolated within 2 h. the amount increased of catalyst to 10 mol% notable significantly with 86 % of **product** and 14 % of **KSM** (Table 1, entry 2). Furthermore amount of catalyst does not affect on yield of the reaction (Table 1, entry 3). other parameters like time and Temperatures does not affect on yield. **4b** and **5** decreased considerable. Increasing temperature from 60 °C does not possible in Ethanol solvent.

Table 1. Optimization of the reaction conditions and EtOH used as a solvent [1].

Entry	Catalyst(Mole%)	Temp(°C)	Time	Yield	Purity	KSM	impurity
1	3	60	240	65%	72	08	20
2	5	60	80	87%	92	05	03
3	7	60	80	85%	89	05	06
4	10	60	80	81%	85	04	08
5	12	60	80	83%	87	05	08
6	15	60	80	84%	88	04	09
7	20	60	80	82%	85	12	03
8	CuCl3	60	240	NR	--	--	--
9	Cu(OAc)	60	240	NR	--	--	--
10	Cu(OTf)2	60	240	NR	--	--	--
11	CuO	60	240	55%	78	16	06

All experiment purity based on First crop of the reaction and all analysis done by TLC w/w concretions method. NR: No reactions

After the optimization of reaction condition and catalyst mole 5% and 4 volume solvent EtOH.

Table 2 optimization of reaction for different derivatives

Entry	Aldehyde	Time	Yield(%)	MP°C	MP°C(Reported)
1	Benzaldehyde	60	84	292	292-294
2	Isobutyraldehyde	80	89	236	235-236
3	4-Chlorobenzaldehyde	70	84	266	265-267
4	trans-Cinnamaldehyde	70	89	202	201-203
5	Hexanal	90	88	163	164-165

In recent years organic chemist focused on green chemistry for organic synthesis [29] because the silica supported catalyst is more stable, efficient and easy to recover and reusable after reaction. The silica supported copper catalyst is design from E-Waste, It formation of benzimidazoles from diamines and aldehydes. it is very known work for check the catalyst activity [30] , we noted in our research in novel method develop for formation of benzimidazoles by the condensation of diamine and aldehydes react via catalytic amount of silica supported copper nano particles under EtOH as reaction media . The reactions done at 60° C temperature for 60 to 90 min by taking a 1:1.2 mol ratio mixture of diamine and aldehydes in the presence of 5 mol% catalyst to give the desired products in and notable (Scheme 1). As shown in Table 2, aromatic, aliphatic, and unsaturated aldehydes use as substituent for reaction react without any difficulty and gave approx same yield and near similar time, difference in rates to give the corresponding 2-substituted benzimidazoles superb yield..

Supportive Spectroscopic data:

Entry 01:

pale yellow solid, FT-IR (KBr) 3450, 3042, 2910, 2100, 1630, 1580, 1470, 1423, 1373, 1202, 1118, 980, 821cm⁻¹ ¹H NMR (400 MHz, DMSO-d₆) : 8.17 (2H), 7.70-7.32 (5H), 7.10-7.00 (1H), 2.47 (3H); ¹³C NMR (100 MHz, DMSO-d₆) :149.9, 142.2, 141.0, 132.3, 134.0, 132.1, 130.7, 125.7, 122.4, 117.2, 110.1, 20.4;. MS (ESI) m/z: [M+H]⁺ calcd :209.10, found 211.

Entry 02:

Greenish solid, FT-IR (KBr):1630, 1580, 1437, 1434, 1320,cm⁻¹ ¹H NMR (400 MHz, DMSO-d₆) : 7.70-7.12 (2H), 7.20-7.00 (2H), 3.24(1H)1.27 (6H); ¹³C NMR (100 MHz, DMSO-d₆) : 141.10, 138.2, 128.7,115.2,29.2, 20.3;. MS (ESI) m/z: [M+H]⁺ calcd :161.10, found 160.2.

Entry 03:

pale yellow solid; FT-IR (KBr) 3441, 2997, 2949,2110, 1626, 1585, 1487, 1422, 1310, 1270, 1220, 1108, 1010cm⁻¹ ¹H NMR (400MHz, DMSO-d₆) :12.90(1H), 8.19 (2H), 7.70 (1H), 7.64 (2H), 7.50 (1H), 7.25-7.20

(2H); ¹³C NMR (100 MHz, DMSO-d₆) :149.1,142.9, 135.1, 131.4, 130.2, 129.7, 128.1, 121.7, 121.2, 118.1, 111.4, MS (ESI) m/z:[M+H]⁺ calcd 229.05, found 230.23.

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